

Flavonoid profile of two antihypertensive extracts of *Trema orientalis* Blume leaves (Cannabaceae)

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ABSTRACT: In order to promote congolese plants with antihypertensive properties, two hypotensive (antihypertensive) extracts of *Trema orientalis* B. leaves were selected in order to know their flavonoid profiles. Therefore, these extracts were prepared and underwent bioguided fractionations until the isolation of the compounds on silica gel. These ethanolic and aqueous extracts presented similar characteristics (appearance, solubility in polar solvents, color of spots at UV 365 nm, frontal ratio). Five compounds were isolated from these extracts of *Trema orientalis* leaves. The analytical data of these compounds show a similarity of physical and chemical characteristics. These compounds presented in NMR¹H the same basic structure of the flavonoid type, 5,7,4'-trihydroxyl-2,3-dihydroflavonol (Kaempferol) or 5,7,4'-trihydroxyl-2,3-dihydroisoflavonol of gross formula C₁₅H₁₂O₆ and differ on their osidic part. Therefore, the definitive structural determination of these compounds must be confirmed by other more in-depth analyses. They will allow to valorize this plant by the production of phytoproducts rich in flavonoids antihypertensive and with oxidizing power.

KEYWORDS: flavonoid profile, trema orientalis, antihypertensive extracts, kaempferol, bioguided fractionation.

1 INTRODUCTION

Despite the damage caused by –harmful microorganisms, the ineffectiveness of some drugs currently available in the treatment of certain diseases makes it essential to search for new and more effective substances to combat all these diseases (cancer, diabetes, high blood pressure, AIDS, etc.) [1], [2], [3], [4], [5], [6].

To solve these problems, the modern pharmaceutical industry now relies on the diversity of metabolites from the plant world to find new molecules with novel biological properties [7], [8], [9], [10], [11], [12]. This source of metabolites seems inexhaustible since a small part of the known plant species have been studied on the phytochemical and biological levels, and each species can contain different bioactive or non-bioactive compounds [1], [13], [14]. These compounds are generally water-soluble such as alkaloids, polyphenols (especially flavonoids), terpenoids and steroids (especially saponins and cardiotonic heterosides) [11], [15], [16], [17]. Many studies suggest that polyphenols, especially flavonoids and tannins, are used as drugs to cure high blood pressure [18], [19], [20], [21].

In this context, isolating and identifying chemical substances with biological properties from plants is a major problem in phytochemistry and/or pharmacognosy. This requires means and techniques that sometimes allow obtaining the desired bioactive compounds [1], [22], [23].

To further enhance and identify substances with the biological properties of these plant species, we focused on the species *Tréma orientalis* Blume (Cannabaceae), a plant used by traditional health in Africa and Asia to treat a large number of pathologies [22], [23], [24], [25], [26], [27], [28], [29], [30].

In Africa in general and in Republic of Congo in particular, some work on extracts (alcoholic and aqueous) of the leaves of this plant has made it possible to evaluate their toxicity and proven the hypotensive (antihypertensive) effects on Wistar rats and vasorelaxant effects on the aortas of these rats [16], [19], [31], [32], [33], [34], [35], [36]. In addition, chemical studies carried out on *Tréma orientalis* B. have made it possible to identify several compounds without biological evaluation [12], [15], [19], [22], [23], [31]. It contains (–)-epicatechin and (+)-catechin, xanthenes and other molecules isolated and identified in its different parts [32], [37], [38], [39], [40].

The importance of polyphenols in medicine prompted us to know the phytochemical profiles of flavonoids present in the antihypertensive extracts of *Trema orientalis* B.

2 MATERIALS AND METHODS

2.1 PREPARATION OF PLANT MATERIAL

Trema orientalis B. leaves were collected in Brazzaville, Republic of Congo. Plant identification was made and recorded (specimen Sita, No. 445) at the National Herbarium of the Centre for Plant Resources Studies in Brazzaville. These leaves were dried in the shade at room temperature for three weeks and then ground to obtain a powder.

2.2 PREPARATION OF EXTRACTS

The aqueous and alcoholic extracts are two hypotensive (antihypertensive) extracts of this plant, consumed by the Congolese population and pharmacologically validated, which were chosen and made the subject of this study. The aqueous extract was obtained after 5 % aqueous decoction of these leaves powder in 30 minutes. The alcoholic extract was obtained after successive exhaustion in Soxhlet of 100 g of powder of these leaves with 500 mL respectively of hexane, chloroform, ethyl acetate and ethanol [19], [23]. The significant mass of each extract was obtained by carrying out these operations several times.

2.3 CHEMICAL SCREENING OF POLYPHENOLS

The screening for metabolites of the extracts was done by Thin Layer Chromatography (TLC) using as support the aluminum plates in silica gel 60 F254 (Merck). The compounds such as flavonoids, phenolic acids and tannins were eluted and revealed using respectively the solvent system AcOEt/HCO₂H/H₂O (80/10/10 V/V/V) and the specific reagents Neu and FeCl₃ at 2 % for respectively flavonoids, phenolic acids, tannins and phenols. The observation of these compounds was made in the visible and UV 365 nm [19], [40].

2.4 TESTS OF CARBONYL GROUPS OF EXTRACTS

These tests were carried out by tube reactions according to the methods described in the literature. The developers used are FeCl₃, AlCl₃ and 2,4-dinitrophenylhydrazone (DNPH) reagents with or without heating [41], [42], [43].

2.5 ISOLATION OF COMPOUNDS ON OPEN COLUMN FROM EXTRACTS

The extract was deposited on a silica gel column (Kieselgel 60 type) prepared in hexane. The elution was carried out respectively by a polarity gradient of the hexane/ethyl acetate system (100/00, 70/30, 50/50, 80/20, 00/100 V/V) followed by another ethyl acetate/ethanol mixture system (90/10, 80/20, 60/40 V/V) to obtain dry fractions after evaporation of the solvents. The last fraction obtained was fractionated again using successively the eluents petroleum ether/ethyl acetate (99/1, 0/100 V/V), ethyl acetate/methanol/formic acid (90/8/2, 85/11/4, 82/14/4, 80/15/5, 77/16/7, 60/40/00 V/V/V) and ethyl acetate/ethanol/water (95/10/5, 76/19/05 V/V/V) until obtaining subfractions each showing a single spot on TLC. These subfractions obtained were purified with the same elution system ethyl acetate/ethanol/water with polarities (85/10/5 V/V/V) and (77/17/6 V/V/V). The powder masses of the fractions or compounds were obtained after evaporation of the solvents in a rotary evaporator at 40 °C. This operation was monitored on TLC at UV 365 nm after revelation of these compounds with Neu reagent [19], [23].

2.6 STRUCTURAL ANALYSIS OF ISOLATED COMPOUNDS

The proton nuclear magnetic resonance ($^1\text{H NMR}$) analyzes were carried out in deuterated dimethyl sulfoxide (DMSO). The NMR spectra of the compounds were obtained and recorded with the NMR devices of the Analysis Pole of Organic Compounds in Poitiers (France).

The processing of the spectra presented in this work was carried out using MesRenova program.

3 RESULTS ET DISCUSSION

Figure 1 shows the chromatographic profiles of the polyphenols of these two extracts of *Trema orientalis* leaves. The chromatographic profiles revealed with Neu and observed in UV 365 nm showed the presence of blue, green and yellow-orange fluorescence spots attributable to phenolic acids and flavonoids (Chromatogram A) [1], [23], [40].

However, we note a dominance of two spots of green and yellow-orange color (frontal ratios 0.29 and 0.38) in UV in the ethanolic and aqueous extracts. These spots observed in visible light take on a yellow and orange color respectively (Chromatogram B).

The revelation of these compounds with the 2 % FeCl_3 solution presented these spots with brown and violet colors in visible light at the same frontal ratios (Chromatogram C). These spots characterize the presence of tannins [1], [23].

Furthermore, the detection of carbonyl groups in these extracts was positive with FeCl_3 , AlCl_3 and DNPH; which proves that these extracts contain compounds having this organic group in their structures. However, the presence of the carbonyl group could disqualify condensed tannins and validate the abundance of flavonoids in these extracts. The reaction of FeCl_3 on these polyphenols does not certify the presence of a carbonyl group in these compounds because this reagent also reacts with phenol [41], [42], [43].

These results confirm the work published in the literature by several authors who have shown the presence of flavonoids, phenolic acids and tannins in polar extracts obtained from the leaves of this plant in several countries of the world [40], [44], [45], [46], [47], [48], [49].

The biological properties of these polyphenolic compounds could justify the use of the leaves of this plant in traditional medicine to treat diseases such as high blood pressure, female sterility, malaria, cough, etc. [8], [15], [18], [44], [46], [50], [51].

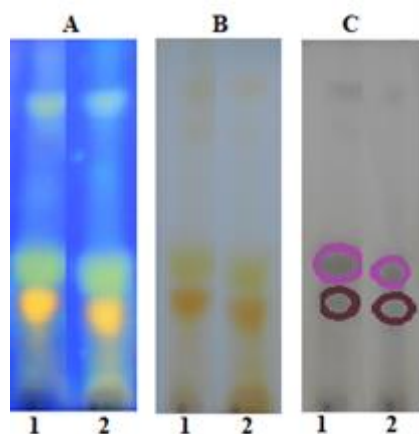


Fig. 1. Chromatographic profiles of polyphenols in extracts of these leaves

1: Ethanolic extract, 2: Aqueous extract, A: Developer: Neu, Observation: Visible; B: Developer Neu and observation of UV 365nm, C: Developer FeCl_3 to 2% and Visible observation.

Fractionation of these two extracts of *Trema orientalis* leaves allowed the isolation of 05 yellow solids including 04 solids noted compounds C1 to C4 with respective yields of 0.007 %; 0.003 %; 0.006 % and 0.005 % relative to 5 g of ethanolic extract and 01 solid noted compound C5 with a yield of 0.015 % relative to 2 g of aqueous extract (Figure 2). After revelation with Neu reagent and observation at UV 365 nm on TLC, these compounds each presented a single green or orange spot, with a frontal ratio of 0.38 in an solvent system ($\text{AcOEt}/\text{HCOOH}/\text{H}_2\text{O}$ 80/10/10, V/V/V), which is characteristic of polyphenols, particularly

flavonoids. These results corroborate and confirm the work of Bonazaba *et al.* who identified spots of the same characteristics, on TLC at UV 365 nm, in the fractions from the ethanolic extract of leaves of the same congolese plant species [19], [23]. The presence of these compounds in this plant has also been confirmed by several works published in the literature [13], [17], [22], [52], [53], [54], [55]. These compounds being soluble in very polar solvents; which made it possible to consider the presence of very polar substituents such as osidic motifs in these compounds and to provide the additional information necessary for structural elucidation [1], [13], [40], [41], [56], [57], [58].

This solubility made it possible to carry out spectral analyses in deuterated DMSO to obtain usable ^1H NMR spectra (Figure 3).

Analysis of these proton spectra revealed that these compounds have a similarity of signals between 04.98 and 13.20 ppm; which allows us to say that they have an identical fragment of their basic structural. These compounds are differentiated by proton signals between 00.00 and 04.98 ppm; these are characteristic of the presence of other chemical groups or radicals associated with this basic fragment. These spectra show that these are polyphenols, in particular flavonoids, containing at least one carbohydrate in their structures [59].

The close similarities are observed between compounds 1 and 2 which exhibit 26 proton signals. However, they are differentiated in ^1H NMR by the presence and absence of a characteristic CH_3 peak ($\delta = 0.46$ ppm) and two signals at 4.39 and 4.47 ppm respectively in these compounds.

Information on compounds 1, 2 and 5 allows one to consider the possibility of isomers or compounds with identical motifs.

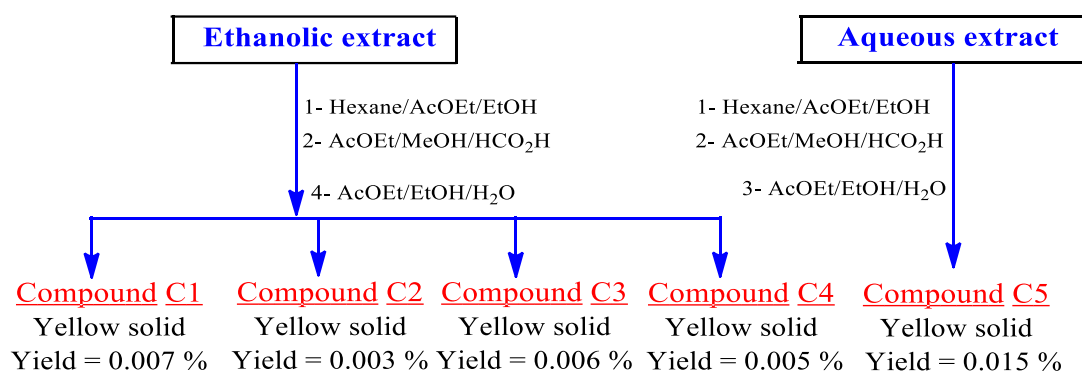


Fig. 2. Scheme of isolation of compounds from *Trema orientalis* B leaves

Considering the similarity of the characteristics (appearance, color, frontal ratio, solubility, NMR and mass spectra) presented by these compounds isolated from these two extracts of the leaves of *Trema orientalis* B., some structural elements of these compounds are proposed from compound C1.

^1H NMR spectrum of compound 1 shows that this molecule has signals at $\delta = 6.26$ (1H, sl), $\delta = 6.80$ (1H, sl), $\delta = 6.90$ (2H, d, $J = 8.80$ Hz) and $\delta = 8.04$ (2H, d, $J = 8.80$ Hz) characteristic of the aromatic protons (H_8 , H_6 , H_3' and H_2') of the basic structure of flavonoids substituted in position 5 and 7 of his part A and in position 4' of his part B (Figures 4 and 5) [23], [59], [60], [61], [62], [63]. Two signals at $\delta = 10.38$ (1H, s) and $\delta = 10.80$ (1H, s) correspond respectively to the protons of the OH function *para*-substituted on the part B and substituted in position 5 on the part A of the flavonoids. Another signal resonating at $\delta = 13.15$ (1H, s) is observed, this is characteristic of an OH function in position 7, in the vicinity of the carbonyl group of the basic ring of the flavonoids. These ^1H NMR data allow us to suppose that these five flavonoids have a basic structure of the 5,7,4'-trihydroxyl-2,3-dihydroflavonol (Kaempferol type) or 5,7,4'-trihydroxyl-2,3-dihydroisoflavonol. Furthermore, signals are observed at $\delta = 0.46$ (3H, d, $J = 6.10$ Hz), between $\delta = 3.22$ and $\delta = 3.57$ (m), at $\delta = 4.32$ (1H, sd, $J = 5.20$ Hz) and at $\delta = 5.26$ (1H, sd, $J = 4.70$ Hz) which could be attributed to carbohydrate or sugar protons (Figures 4 and 6) [23], [63], [64].

The definitive structural determination of these compounds must be confirmed by further, more in-depth analyses.

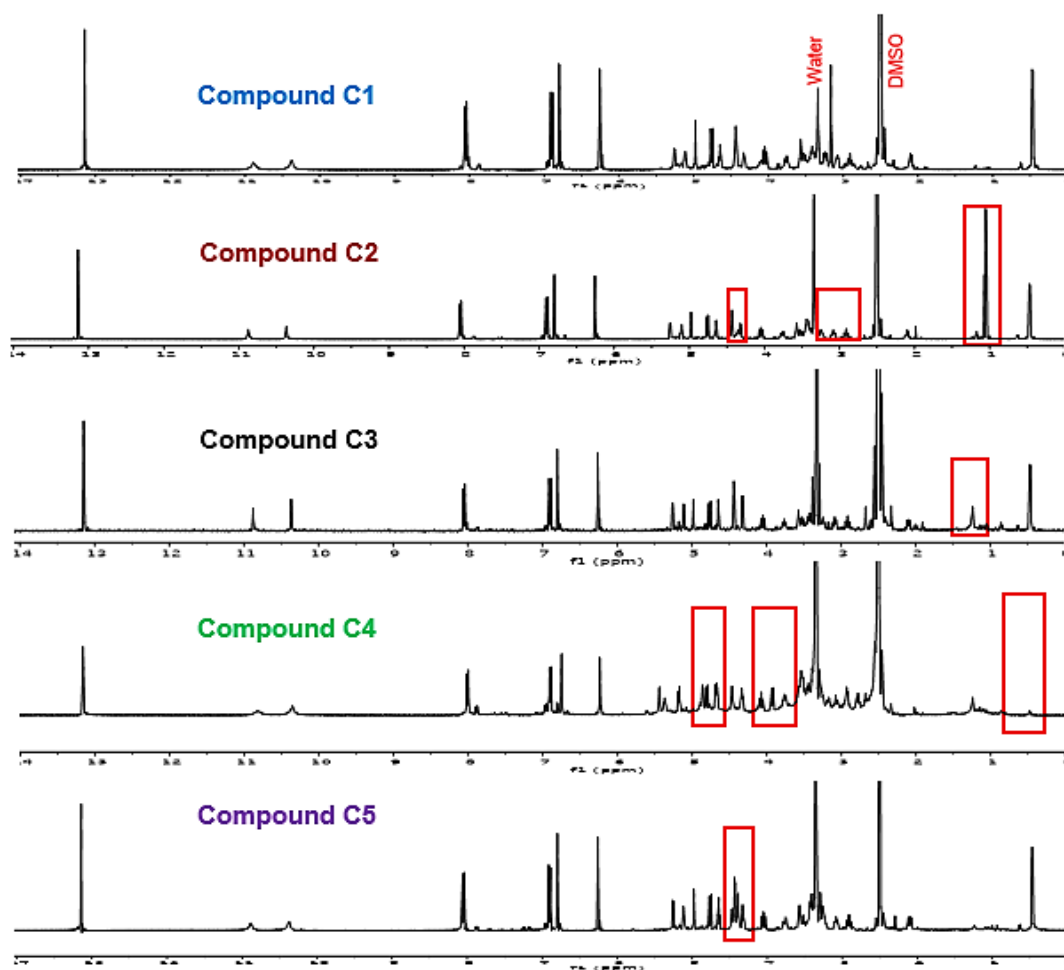


Fig. 3. ^1H NMR spectra of compounds isolated from *Trema orientalis* B. leaves and carried out in deuterated DMSO

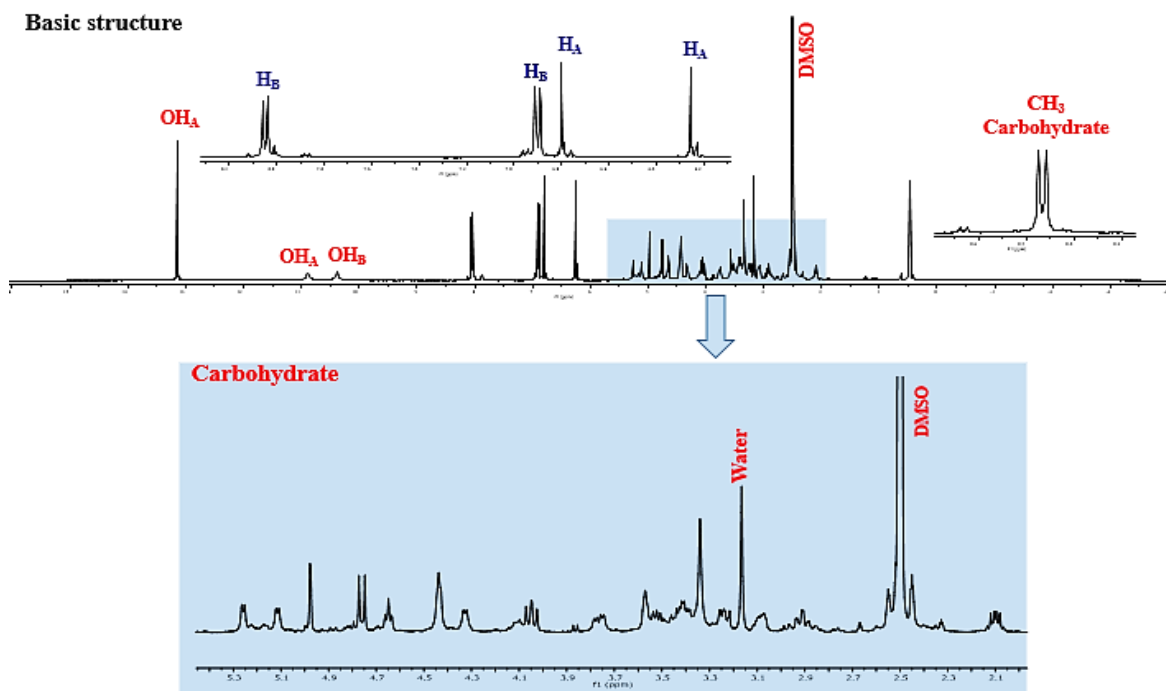


Fig. 4. Detailed ^1H NMR spectra of compound C1 isolated of *Trema orientalis* B. leaves and performed in deuterated DMSO

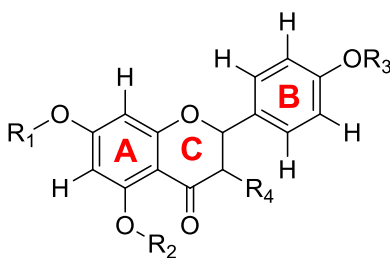


Fig. 5. Supposed basic structure of the compounds

R_1 at R_4 : H ou Carbohydrate



Fig. 6. Structural elements of carbohydrates identified in the isolated compounds

4 CONCLUSION

This work presents a phytochemical study to identify flavonoids from *Trema orientalis* leaves, an antihypertensive plant used in traditional congolese medicine. Among these compounds, five compounds were isolated from this plant. These compounds presented similar characteristics (appearance, solubility in polar solvents, color of spots at UV 365 nm, frontal ratio). The information on the proton spectra allowed to suppose the presence of a basic chemical structure of the type of substituted flavonoids (5,7,4'-trihydroxyl-2,3-dihydroflavonol) or 5,7,4'-trihydroxyl-2,3-dihydroisoflavonol) of at least one sugar in these compounds. This information requires in the future a definitive identification and confirmation of these flavonoids by other more in-depth analyses. They will allow to valorize this plant by the production of antihypertensive phytoproducts rich in flavonoids.

DECLARATION OF INTEREST

The authors declare that they have no conflict with any third party.

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